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(54) Title: METHODS FOR PREPARING OF WEATHERABILITY THERMOPLASTIC RESIN

(57) Abstract: The present invention relates to a method for preparing a weatherable thermoplastic resin, and more particularly to a method for preparing a thermoplastic acrylate-styrene-acrylonitrile (ASA) resin having superior weatherability and colorability. In detail, the present invention provides a method comprising the steps of: a) preparing a seed by adding a water-soluble initiator to an aromatic vinyl compound, or to a (meth)acrylic acid alkyl ester, and polymerizing; b) preparing an alkyl acrylate rubber polymer by adding an alkyl acrylate monomer, an emulsifier of pH 3~9, and an oil-soluble initiator to the seed, and polymerizing; and c) preparing a thermoplastic resin by adding an aromatic vinyl compound, a vinyl cyanide compound, an emulsifier of pH 9~13, and an oil-soluble initiator to the alkyl acrylate rubber polymer, and graft polymerizing.

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METHODS FOR PREPARING OF WEATHERABLILITY

THERMOPLASTIC RESIN

BACKGROUND OF THE INVENTION

(a) Field of the Invention

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The present invention relates to a method for preparing thermoplastic resin having weatherability, and more particularly to a method for preparing thermoplastic acrylate-styrene-acrylonitrile (ASA) resin having excellent weatherability and color matching properties.

(b) Description of the Related Art

Recently, as the thermoplastic resin industry has advanced and the distinctiveness of goods model has gained popularity, research on giving high function of colorability etc, to materials used in the industry has progressed a great deal.

In order to increase resin functionality, the prior art has used an acrylonitrile-butadiene-styrene (ABS) copolymer resin. Although ABS resin has good qualities of impact resistance, processibility, and surface gloss, etc., it has a problem in that it is unsuitable for use outdoors due to bad weatherability and light resistance because of properties of the resin, and thermal stability is also bad, chemical resistance is weak, and its properties diminish over time.

In order to solve these problems, acrylate-styrene-acrylonitrile (ASA) resin is being developed.

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In a method for preparing ASA resin of the prior art, Japanese Laid-Open Patent Publication No. Hei 5-202264 has suggested a method for improving properties by preparing ASA resin with a bimodal shape. The method discloses that ASA resin with a small diameter of 50 ~ 200 nm, and ASA resin with a large diameter of 200 ~ 1000 nm each prepares, and then the two resins are latex blended together, and resultant blends with a separately prepared styrene-acrylonitrile (SAN) copolymer to produce ASA resin having superior weatherability.

However, since the method separately prepares two ASA resins the process is complex, and colorability, surface gloss, and sufficient impact strength are not exhibited. German Patent No. 1260135 has described the method in detail.

In another method, Japanese Laid-Open Patent Publication No. Hei 4-180949 discloses a method for preparing ASA resin by making multi-layer graft copolymer particles. The method suggests a preparation of ASA resin for improving weatherabilty in which a hard core is first prepared by using a monomer having a high glass transition temperature, then a crosslinked acrylate rubber component is added, followed by a crosslinked shell of a styrene monomer, an acrylonitrile monomer having a high glass transition temperature, and a crosslinking agent, and finally by preparing a soft shell with an uncrosslinked styrene monomer and an acrylonitrile. However, this method also does not

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mention a method capable of improving colorability and surface gloss of ASA resin.

As explained, although a general ASA resin has weatherability, chemical resistance, and thermal stability, it has problems of a lowering colorability and surface gloss, weak side impact strength, and its molding capacity is limited due to a low fluidity. Additionally, important technical difficulties to overcome in the preparation of ASA resin are known to be that colorability and surface gloss showing a pearl color etc. while color matching are reduced, and impact strength is low. Also, color dispersion falls due to properties of acrylate rubber.

SUMMARY OF THE INVENTION

The present invention is made in consideration of the problems of the prior art, and it is an object of the present invention to provide a method for preparing thermoplastic acrylate-styrene-acrylonitrile (ASA) resin having superior weatherability, as well as properties and colorability equal to ABS resin.

It is another object of the present invention to provide a method for preparing thermoplastic resin having superior impact resistance, chemical resistance, and surface gloss.

In order to achieve these objects, the present invention provides a method for preparing a weatherable thermoplastic resin comprising the steps of:

a) preparing a seed by adding 0.05-0.2 weight parts of a water-soluble initiator to 0.5-10 weight parts of either an aromatic vinyl compound or a

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(meth)acrylic acid alkyl ester,

and polymerizing;

- b) preparing an alkyl acrylate rubber polymer by adding
 - i) 5-45 weight parts of an alkyl acrylate monomer,
- 5 ii) 1-5 weight parts of an aromatic vinyl compound, a (meth)acrylic acid alkyl ester, or a vinyl cyanide compound,
 - iii) 0.2-1.0 weight parts of a metal salt derivative of alkyl sulfosuccinate with $C_{12} \sim C_{18}$, or a metal salt derivative of alkyl sulfuric acid ester, or metal salts of sulfonate, with $C_{12} \sim C_{20}$, as an emulsifier, and
- iv) 0.05-2 weight parts of an oil-soluble initiator to the seed,
 - and polymerizing; and
 - c) preparing a thermoplastic resin by adding
 - i) 10-40 weight parts of an aromatic vinyl compound,
- ii) 1-20 weight parts of a vinyl cyanide compound,
 - iii) 0.5-3.0 weight parts of metal salts of rosinate or metal salts of carboxylate, with $C_{12} \sim C_{20}$, as an emulsifier, and
 - iv) 0.05-0.2 weight parts of an oil-soluble initiator
 to the alkyl acrylate rubber polymer, and graft copolymerizing.
- The present invention also provides a weatherable thermoplastic resin prepared by the above method.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

The present invention will now be explained in detail.

In order to solve the problem of the method for preparing weatherable resin that lowers colorability and surface gloss of acrylate-styrene-acrylonitrile (ASA) resin of the prior art, the present inventors continuously studied. As a result, the present invention was completed by introducing methylmethacrylate, by controlling contents of each component, by introducing a hard core during polyacrylate rubber polymerization, and by using a water-soluble initiator and an oil-soluble initiator differently in the polymerization steps to prepare ASA resin. The resin of the present invention comprises an acrylonitrile giving superior chemical resistance, an acrylate-based rubber giving superior impact resistance, and a styrene giving superior processability, to prepare a thermoplastic resin having superior impact resistance, chemical resistance, processability, weatherability, colorability, and surface gloss.

The method for preparing thermoplastic resin having superior weatherability of the present invention comprises the steps of a) preparation of a seed, b) preparation of a crosslinked alkyl acrylate rubber polymer, and c) preparation of a graft copolymer, and each process is explained in more detail as follows.

a) Preparation of a seed

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First, in order to produce a hard core for preparing a crosslinked alkyl acrylate rubber polymer, the present invention prevents distortion of the rubber by hardening a rubber core through seed polymerization of 0.5 to 10 weight parts of an aromatic vinyl compound monomer or a (meth)acrylic acid alkyl ester monomer having a high glass transition temperature, based on 100 weight parts of the total monomers.

Examples of preferable (meth)acrylic acid alkyl ester compounds are (meth)acrylic acid methyl ester, (meth)acrylic acid ethyl ester, (meth)acrylic acid propyl ester, (meth)acrylic acid 2-ethyl hexyl ester, (meth)acrylic acid decyl ester, and (meth)acrylic acid lauryl ester etc., and more preferable is a methyl methacrylate such as (meth)acrylic acid methyl ester.

The aromatic vinyl compound used is a styrene monomer derivative.

Preferable examples are selected from the group consisting of styrene, a-methyl styrene, and vinyl toluene, and particularly preferable is styrene.

According to the present invention, in the case of initial seed polymerization, a water-soluble initiator is used; and in the case of polymerization in second and third steps, an oil-soluble initiator is used, and thus colorability and surface gloss can be improved by increasing graft efficiency of the reactant. The initiator used in the present invention is an inorganic or an organic peroxide. In the case of seed polymerization, the water-soluble initiator is used to improve colorability, and the oil-soluble initiator is used in the graft and rubber particle

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growth reaction. As examples, the water-soluble initiator is preferably selected from the group consisting of potassium persulfate, sodium persulfate, and ammonium persulfate, while the oil-soluble initiator is preferably cumene hydroperoxide, t-butyl hydroperoxide, or benzoyl peroxide.

The content of the water-soluble initiator is preferably 0.05 to 0.2 weight parts based on 100 weight parts of total monomers.

The crosslinking agent used in the present invention is selected from the group consisting of ethyleneglycol dimethacrylate, diethyleneglycol dimethacrylate, triethyleneglycol dimethacrylate, 1,3-buthanediol dimethacrylate, 1,6-hexanediol dimethacrylate, neopentylglycol dimethacrylate, trimethylol propane trimethacrylate, and trimethylol methane triarylate. The content is 0.05 to 0.3 weight parts based on 100 weight parts of total monomers.

An electrolyte used in the present invention is selected from the group consisting of $NaHCO_3$, $Na_2S_2O_7$, K_2CO_3 , and preferable is $NaHCO_3$. The content of electrolyte is 0.05 to 0.4 weight parts based on 100 weight parts of total monomers.

b) Preparation of crosslinked alkyl acrylate rubber polymer

The present invention performs the step of preparing a crosslinked alkyl acrylate rubber polymer by adding an alkyl acrylate monomer; an aromatic vinyl compound, a (meth)acrylic acid alkyl ester, or a vinyl cyanide compound; a metal salt derivative of alkyl sulfosuccinate with $C_{12} \sim C_{18}$, or a metal salt derivative of

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alkyl sulfuric acid ester or metal salts of sulfonate with $C_{12} \sim C_{20}$, as an emulsifier; and the oil-soluble initiator, to the seed prepared in the a) step.

In the present invention, an alky acrylate monomer (preferably butyl acrylate) as a major monomer is used, and the content thereof is 5 to 45 weight parts based on 100 weight parts of total monomers.

Additionally, the present invention preferably uses 0.1 to 15 weight parts of an optimized mixture ratio of methyl methacrylate, styrene, and acrylonitrile as a supplement to improve efficiency of the rubber particles and the pearl color, and it is also possible to copolymerize a functional monomer. The functional monomer is selected from the group consisting of methacrylic acid, acrylic acid, maleic acid, itaconic acid, and fumalic acid.

An aqueous solution of the emulsifier preferably has pH of 3 \sim 9, and for example the emulsifier used is a metal salt derivative of alkyl sulfosuccinate with $C_{12}\sim C_{18}$, or a derivative of alkyl sulfuric acid ester or metal salts of sulfonate with $C_{12}\sim C_{20}$. The metal salt derivative of alkyl sulfosuccinate with $C_{12}\sim C_{18}$ is a sodium or potassium salt etc. of dioctyl sulfosuccinate, dicyclohexyl sulfonate, or dihexyl sulfosuccinate; and the derivative of alkyl sulfonic ester with $C_{12}\sim C_{20}$ or metal salts of sulfonate can be used as alkyl sulfate metal salts such as sodium lauric sulfate, sodium dodecyl sulfate, sodium dodecyl benzene sulfate, sodium octadecyl sulfate, sodium oleic sulfate, potassium dodecyl sulfate, potassium octadecyl sulfate, etc. Among them, in overall consideration of the graft reaction

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of the graft phase for foam occurrence and the drying property of the crosslinked butyl acrylate rubber polymer, sodium dioctyl sulfosuccinate or potassium dioctyl sulfosuccinate is preferable. The content of emulsifier used is 0.2 to 1.0 weight parts based on 100 weight parts of total monomers.

If a carboxylate metal salt derivative such as a metal salt of aliphatic acid or a metal salt of rosiniate etc. where an aqueous solution of the emulsifier has pH of about 9 ~ 13 is used, although drying can be enhanced, it is difficult to graft methacrylic acid methyl in the butyl acrylate rubber polymer into ASA resin due to a lower moisture content, and thus stripping occurs because of phase separation resulting in a substantial decrease in impact strength.

The oil-soluble initiator used in the present invention is currene hydroperoxide, t-butyl hydroperoxide, or benzoyl peroxide. The content is 0.05 to 0.2 weight parts based on 100 weight parts of total monomers.

Also, according to the preparation of the acrylate rubber polymer of the present invention, the b) step can further comprise a grafting agent selected from the group consisting of v) aryl methacrylate (AMA), triarylisocyanurate (TAIC), triarylamine (TAA), and diarylamine (DAA), wherein the content is 0.01 to 0.07 weight parts based on 100 weight parts of total monomers.

The crosslinking agent used in the present invention is selected from the group consisting of ethyleneglylcol dimethacrylate, diethyleneglycol dimethacrylate, triethyleneglycol dimethacrylate, 1,3-buthanediol dimethacrylate,

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1,6-hexanediol dimethacrylate, neopentylglycol dimethacrylate, trimethylol propane trimethacrylate, and trimethylol methane triarylate. The content is 0.05 to 0.3 weight parts based on 100 weight parts of total monomers.

The electrolyte used in the present invention is selected from the group consisting of NaHCO₃, Na₂S₂O₇, and K₂CO₃, and NaHCO₃ is preferable. The content of electrolyte is 0.05 to 0.4 weight parts based on 100 weight parts of total monomers.

The reaction can be performed only through emulsion polymerization, or it can be preformed by suitably mixing emulsifier-free emulsion polymerization and emulsion polymerization to prepare acryl rubber polymer latex. The monomers can be added continuously, or by adding continuously and in batches.

The aqueous solution of the crosslinked alkyl acrylate rubber polymer latex after polymerization of the present invention preferably has pH of 5 to 9, and more preferably 6 to 8. The rubber polymer preferably has a diameter of 2500 \sim 5000 Å, and more preferably 3000 \sim 4500 Å.

c) Preparation of graft copolymer

The present invention can have improved impact strength and it can maintain colorability of ASA resin better when compared to the prior art through the graft reaction by adding an aromatic vinyl compound, a vinyl cyanide compound, metal salts of rosinate or metal salts of carboxylate with $C_{12} \sim C_{20}$ as an emulsifier, and an oil-soluble initiator to the acrylate rubber polymer prepared

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in the b) step.

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The monomer grafted in the crosslinked alkyl acrylate rubber polymer uses 10 to 40 weight parts of the aromatic vinyl compound and 1 to 20 weight parts of the vinyl cyanide compound, and a functional monomer is further used according to necessity.

The aromatic vinyl compound used is a styrene monomer derivative. As an example, it is selected from the group consisting of styrene, α-methyl styrene, and p-methyl styrene, and preferable is styrene. The vinyl cyanide used is acrylonitrile or methacrylonitrile, and preferable is acrylonitrile. The functional monomer possibly used can be methacrylic acid, acrylic acid, maleic acid, itaconic acid, furmalic acid, etc.

The aqueous solution of the emulsifier preferably has pH of 9 \sim 13, and as an example the emulsifier is preferably a metal salt derivative of carboxylate such as a metal salt derivative of aliphatic acid with $C_{12}\sim C_{20}$, a metal salt of rosinate, etc. The metal salt derivative of aliphatic acid is a sodium or potassium salt of a fatty acid, lauric acid, or oleic acid, and the metal salt of rosinate is sodium rosinate or potassium rosinate.

If a metal salt derivative of carboxylate such as a metal salt of aliphatic acid with $C_{12} - C_{20}$ or a metal salt of rosinate etc. where the aqueous solution emulsifier has a pH of about 9 \sim 13 is used, as the pH of the total system rises little by little, a carboxyl group of butyl acrylate rubber polymer having a low glass

transition temperature comes out of the particle. But as polymerization progresses, a graft phase having a high glass transition temperature comes to the inner part of the butyl acrylate rubber polymer particle, and thus a surface glass transition temperature can be lowed and thereby it can be coagulated at normal pressure, and additionally, drying also becomes easier because the moisture content inside the ASA particle is remarkably decreased.

In the case a graft reaction initiator is used as explained, an oil-soluble initiator such as cumene hydroperoxide, t-butyl hydroperoxide, or benzoyl peroxide is used, and the content is preferably 0.05 to 0.3 weight parts based on 100 weight parts of total monomers.

In the case of the grafting reaction, the mixing monomer comprising an emulsifier is preferably continuously added. With the batch adding method, since pH of the polymerization system rises at the same time, it is difficult to graft and the inner structure is not uniform due to inferior particle stability.

The final latex aqueous solution prepared by the method preferably has pH of 8 \sim 11, and more preferably 9 \sim 10.5. Additionally, a diameter of the rubber polymer latex is preferably 3000 \sim 6000 Å, and more preferably 3500 \sim 4500 Å.

As above, according to the present invention, a weatherable thermoplastic resin which has superior impact resistance, chemical resistance, processability etc., and improved colorability and surface gloss can be obtained

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by controlling an amounts of the monomer components such as a (meth)acrylic acid alkyl ester compound, an aromatic vinyl compound, a vinyl cyanide compound etc. added in a polyacrylate-based rubber latex, and particularly by graft polymerizing them together with a water-soluble initiator and an oil-soluble initiator.

The present invention will be explained in more detail with reference to the following Examples. However, these are to illustrate the present invention, and the present invention is not limited to them.

[Examples]

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10 <u>First polymerization step</u>

ion exchange water

70 wt parts

styrene (SM)

2 wt parts

ethyleneglycol dimethacrylate (EDMA) 0.02 wt parts

sodium bicarbonate

0.1 wt parts

potassium persulfate (KPS)

0.01 wt parts

Second polymerization step

20 potassium rosinate

45 wt parts

dioctylsulfosuccinate

0.5 wt parts

Butylacrylate 40 wt parts
styrene (SM) 3 wt parts
methylmethacrylate (MMA) 2 wt parts
ethyleneglycol dimethacrylate (EDMA) 0.12 wt parts

arylmethacrylate 0.04 wt parts
sodium bicarbonate 0.1 wt parts
cumene hydroperoxide (CHP) 0.06 wt parts

All the above components except cumene hydroperoxide were mixed in a polymerization reactor, and then the obtained mixture and a catalyst were each continuously introduced to the product of the first polymerization step at 70 °C over 4 hours to conduct the polymerization reaction. The diameter of the obtained latex was 4000 Å, the pH thereof was 8, and a polymerization conversion rate was 98%. The diameter of the latex was measured by the dynamic laser-light scattering method using a Nicomp analyzer (Model: 370HPL).

15 <u>Third polymerization step</u>

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ion exchange water 63 weight parts

potassium rosinate 0.5 weight parts

styrene(SM) 38 weight parts

acrylonitrile(AN) 15 weight parts

cumene hydroperoxide(CHP) 0.11 weight parts

All the above components except cumene hydroperoxide were mixed in a

polymerization reactor, and then the obtained mixture and a catalyst were each continuously introduced to the product of the second polymerization step at 70 °C over 4 hours to conduct a polymerization reaction. Thereafter, in order to improve the polymerization conversion rate, the resultant was further reacted at 80 °C for 1 hour, and it was then cooled to 60 °C. The diameter of the obtained latex was 4800 Å, the pH thereof was 9.5, the polymerization conversion rate was 99%, and the graft rate was 45%.

The obtained latex was coagulated at 85 °C, normal pressure with a calcium chloride aqueous solution, then aged at 95 °C, then dehydrated and washed, and finally dried by hot blast at 90 °C for 30 min. to obtain ASA powder particles having weatherability.

To the ASA powder particles having weatherability obtained by the method, a SAN (81HF) copolymer resin (LG CHEM. Ltd. product) was introduced as an antioxidant and a photostabilizer, and the whole was kneaded and extruded at 200 to 250 °C to prepare a pellet having a rubber content of 20 %. The pellet was injection molded again and the properties were measured, and the results are shown in Table 1.

The properties were measured by the ASTM method, and measurement conditions of each property were as follows.

lzod impact strength: ASTM D256 (1/4" notched at 25 ℃, unit: Kg·cm/cm)

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colorability: decision made with the naked eye

surface gloss: ASTM D528 (45° angle)

[Example 2]

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ASA resin was prepared by the same method as in Example 1, except that the amount of styrene used in the first polymerization step was increased to 5 wt parts, and the amount of styrene used in the third polymerization step was reduced to 35 wt parts. The properties were measured by the same method as in Example 1, and the results are shown in Table 1.

[Example 3]

10 ASA resin was prepared by the same method as in Example 1, except that TBHP (t-butyl hydroperoxide was used instead of cumene hydroperoxide as the initiator in the second polymerization step. The properties were measured by the same method as in Example 1, and the results are shown in Table 1.

[Example 4]

ASA resin was prepared by the same method as in Example 1, except that methylmethacrylate was used instead of the styrene monomer in the first polymerization step. The properties were measured by the same method as in Example 1, and the results are shown in Table 1.

[Example 5]

ASA resin was prepared by the same method as in Example 1, except that acrylonitrile was used instead of the styrene monomer in the second

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polymerization step. The properties were measured by the same method as in Example 1, and the results are shown in Table 1.

[Comparative Example 1]

ASA resin was prepared by the same method as in Example 1, except that methylmethacrylate and the styrene monomer were not added in the second polymerization step, and the amounts thereof were added in the third step polymerization. The properties were measured by the same method as in Example 1, and the results are shown in Table 1.

[Comparative Example 2]

ASA resin was prepared by the same method as in Example 1, except that potassium persulfate (KPS) as a water-soluble initiator was used instead of cumene hydroperoxide (CHP) in the second and third polymerization steps. The properties were measured by the same method as in Example 1, and the results are shown in Table 1.

15 [Table 1]

	Exampl	Exampl	Exampl	Exampl	Exampl	Com.	Com.
	e 1	e 2	e 3	e 4	e 5	Exam. 1	Exam. 2
Impact strength (Kg · cm/cm)	20	16	19	20	18	12	10
Surface gloss (45°)	99	101	95	93	95	80	75
Colorability	good	good	good	good	good	Bad	Very bad

As shown in Table 1, Examples 1 to 5 using both a water-soluble initiator and an oil-soluble initiator of the present invention had excellent impact strength at 16 ~ 20 Kg · cm/cm, and surface gloss and colorability were very excellent. On the other hand, Comparative Examples 1 and 2 used only a water-soluble initiator, and the thus-produced ASA resin had a very low impact strength, and colorability was extremely bad.

As explained, the method of preparing a thermoplastic resin of the present invention results in a resin with superior impact resistance, chemical resistance, and processability, etc., and particularly a thermoplastic resin with good weatherability and having improved colorability and surface gloss can be obtained.

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WHAT IS CLAIMED IS:

 A method for preparing a weatherable thermoplastic resin comprising the steps of:

a) preparing a seed by adding 0.05-0.2 weight parts of a water-soluble
 initiator to 0.5-10 weight parts of either an aromatic vinyl compound or a (meth)acrylic acid alkyl ester,

and polymerizing;

- b) preparing an alkyl acrylate rubber polymer by adding
 - i) 5-45 weight parts of an alkyl acrylate monomer,
- ii) 1-5 weight parts of an aromatic vinyl compound, a (meth)acrylic acidalkyl ester, or a vinyl cyanide compound,
 - iii) 0.2-1.0 weight parts of a metal salt derivative of alkyl sulfosuccinate with $C_{12} \sim C_{18}$, or a metal salt derivative of alkyl sulfuric acid ester or metal salts of sulfonate with $C_{12} \sim C_{20}$, as an emulsifier, and
- iv) 0.05-2 weight parts of an oil-soluble initiator to the seed, and polymerizing; and
 - c) preparing a thermoplastic resin by adding
 - i) 10-40 weight parts of an aromatic vinyl compound,
 - ii) 1-20 weight parts of a vinyl cyanide compound,
- 20 iii) 0.5-3.0 weight parts of metal salts of rosinate or metal salts of carboxylate with C₁₂~C₂₀, as an emulsifier, and

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iv) 0.05-0.2 weight parts of an oil-soluble initiator

to the alkyl acrylate rubber polymer, and graft copolymerizing.

2. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the alkyl acrylate rubber polymer obtainable in step b) has a diameter of 2500 \sim 5000 Å and a polymer aqueous solution thereof has pH of 5 \sim 9; and wherein the graft copolymer obtainable in step c) has a particle size of 3000 \sim 6000 Å and an aqueous solution thereof has pH of 8 \sim 11.

- 3. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein an aqueous solution of the emulsifier of step b) has pH of 3 ~ 9, and wherein an aqueous solution of the emulsifier of step c) iii) has pH of 9 ~ 13.
- 4. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the water-soluble initiator of step a) is selected from the group consisting of potassium persulfate, sodium persulfate, and ammonium persulfate.
- 5. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the oil-soluble initiator of step b) and step c) is cumene hydroperoxide, t-butyl hydroperoxide, or benzoyl peroxide.
- 6. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the alkyl acrylate monomer of step b) is selected from the group consisting of butyl acrylate, ethyl acrylate, and methyl acrylate.
 - 7. The method for preparing a weatherable thermoplastic resin according

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to Claim 1, wherein the aromatic vinyl compound of step b) is selected from the group consisting of styrene, α-methyl styrene, and vinyl toluene.

- 8. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the vinyl cyanide compound of step b) is acrylonitrile or methacrylonitrile.
- 9. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the (meth)acrylic acid alkyl ester of step b) is selected from the group consisting of (meth)acrylic acid methyl ester, (meth)acrylic acid ethyl ester, (meth)acrylic acid propyl ester, (meth)acrylic acid 2-ethyl hexyl ester, (meth)acrylic acid decyl ester, and (meth)acrylic acid lauryl ester.
- 10. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the polymerization reaction of step b) is performed while further comprising 0.01-0.07 weight parts of a grafting agent selected from the group consisting of v) aryl methacrylate (AMA), triarylisocyanurate (TAIC), triarylamine (TAA); and diarylamine (DAA).
- 11. The method for preparing a weatherable thermoplastic resin according to Claim 1, wherein the polymerization reactions of step a) and step b) are each performed while further comprising a crosslinking agent selected from the group consisting of ethyleneglylcol dimethacrylate, diethyleneglycol dimethacrylate, triethyleneglycol dimethacrylate, 1,3-buthanediol dimethacrylate, 1,6-hexanediol dimethacrylate, neopentylglycol dimethacrylate, trimethylol

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propane trimethacrylate, and trimethylol methane triarylate.

12. A weatherable thermoplastic resin prepared by the method of Claim 1.

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INTERNATIONAL SEARCH REPORT

International application No. PCT/KR01/02274

A. CLAS	A. CLASSIFICATION OF SUBJECT MATTER						
IPC7	7 C08F 265/06						
According to	According to International Patent Classification (IPC) or to both national classification and IPC						
	numentation searched (classification system followed by	y classification symbols)					
IPC7 C08F		• •					
Documentatio	n searched other than minimum documentation to the c	extent that such documents are included in the	ields searched				
Electronic date	a base consulted during the intertnational search (name	of data base and, where practicable, search ten	ms used)				
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT						
Category*	Citation of document, with indication, where app	propriate, of the relevant passages	Relevant to claim No.				
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